Physical and Chemical Drivers of Deuteration in Protoplanetary Disks

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Fractionation if Isotopes in Space, Florence, Italy

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I. Dense Molecular Cloud ~ 0.5-3 Myr

~ 3-10 Myr III. Protoplanetary Disk

Phases of Star Formation

 $\sim 10^5 \text{ yr}$

II. Protostar

IV. Planetary Systems

>10 Myr

Credit: Bill Paxton



Plot references collected in Cleeves et al. 2014

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2 Bulk **Dense ISM**

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1 Bulk Dense ISM

Primordial ices in the envelopes of protostars exhibit a high level of D/H.

Are these early stages (the primordial ISM ices) chemically linked? What is the role of disk processing?

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ISM: Persson+2014, 2012, Coutens+2012, Parise 2003.

Talk Overview

1) Key Ingredients For Disk Deuterium Chemistry

2) What Are The Disk Initial Conditions?

3) Effect Of Gas Viscous Evolution, Turbulence, And Mixing

4) The Late Phase Redistribution Of Ices

5) Not All Roads Lead To Rome: Variations In Fractionation Pathways

6) Moving Towards A Comprehensive Picture Of Disk Deuterium Chemistry

1) Key ingredients for disk deuterium chemistry



Supports fractionation up to T < 50 K



1) Key Ingredients for High Molecular D/H Ionization, HD H_2 , HD H₃+ H_2D^+ H_2 $H_{2}, \Delta E$ electron, grain H atom Low temperatures such that ΔE becomes

important.





CO, N₂, O, OH, etc.

H, D atoms

lce formation

HCO+

DCO¹

Reactants available to form deuterated molecules.

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 N_2H

 N_2D^+

 H_2O

HDO



Carbon pathways support fractionation up to T < 100 K, see Roueff et al. 2013

Ingredient 1: Ionizing Processes

Radionuclides



CR-dominated

X-ray dominated



Ingredient 2: Disk Thermal Structure

$\mathrm{H}_3^+ + \mathrm{HD} \rightleftharpoons \mathrm{H}_2\mathrm{D}^+ + \mathrm{H}_2 + \Delta E_1$

~10 AU



Temperature + Ionization





X-ray dominated

UV-dominated

W-dominated X-ray dominated

Cool (<50 K) regions of the disk are also in the region most sensitive to CRs.

Ingredient 3: Disk Molecular Structure



Need reactive species to transfer D to, CO, N₂, O, ...

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 $H_3^+ H_2D^+$

DCO*

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Hot-Phase Deuterium Chemistry

The hot inner disk also can fractionate through neutral neutral isotopic exchange.
e.g., H₂O + HD ↔ HDO + H₂
D + OH ↔ H + OD (e.g. Thi et al 2010)

Yang et al. 2013



e.g., Drouart 1999, Mousis 2000, Hersant et al. 2001, Thi et al 2010

2) Disk Initial Conditions?









Yang and Ciesla 2012

* Mass of the envelope? Initial angular momentum?

* Thermal history of infalling material? Shocks? Is it symmetric or streamers?

* Subsequent viscous +disk+stellar evolution?

2



Visser, van Dishoeck, Doty & Dullemond 2009







Herbst & van Dishoeck 2009

2) What are the initial disk chemical conditions? Cloud Core H₂O Fractional composition, D/H **10**⁻¹ \odot **10**⁻² H,CO **10**⁻³ **CH**₂OH D_2O/H_2O 10⁻⁴ HDO/H₂O CO & CH₃OH-rich 10⁻⁵ (layer II) H₂O-dominated 20 30 50 60 70 80 10 40 H₂O-dominated (layer I) (layer I) Num. of the total ice layers dust dust $f_{D2} < f_{D1} < 10^{-3}$ $f_{D2} < f_{D1} < 10^{-3}$ $10^{-3} << f_{D2} < f_{D1}$ Furuya, Aikawa et al. 2015, and Furuya, van Dishoeck, & Aikawa 2015

By modeling D₂O/HDO, HDO/H₂O and using layered ice model the Furuya +2015 models can explain water's early evolution.





Initial Cloud Angular Momentum is Key

Yang and Ciesla 2012, Yang, Ciesla, Alexander 2013

Low angular momentum





Initial Cloud Angular Momentum is Key Yang and Ciesla 2012, Yang, Ciesla, Alexander 2013

0.3 Myr

Low angular momentum





Initial Cloud Angular Momentum is Key Yang and Ciesla 2012, Yang, Ciesla, Alexander 2013

High angular momentum





Initial Cloud Angular Momentum is Key

Yang and Ciesla 2012, Yang, Ciesla, Alexander 2013



High angular momentum

Rcf ~40 AU



* w_{cloud} = 10⁻¹⁴ s⁻¹ typical is 10⁻¹⁵ - 10⁻¹³ s⁻¹





Yang, Ciesla, Alexander 2013

2) What are the initial disk chemical conditions? Violent disk processing through young stellar outbursts?





Owen and Jacquet 2015

3) Impact of Gas Disk Evolution

Post protostellar gas kinematics



Keplerian Motion
 + Thermal
 Pressure Support

• Turbulence?

Viscous
 Evolution/
 Accretion?

Winds?
 Photoevap. or
 MHD?

3) The role of mixing/turbulent evolution?

* Mixing: The jury is still out.

* Theory demonstrates mixing can both produce significant D/H enhancements in water and decrease D/H (Albertsson et al. 2015 and Furuya et al. 2013).

* May also operate on other deuterated species, including organics.




r, AU

'Turbulent mixing slowly transports some of the water ice into warmer or irradiated regions where it desorbs and is quickly defractionated by ion-molecule and dissociative recombination processes in the gas phase." - Albertsson et al. 2014



"Still, transport of oxygen affects water and deuterated water chemistry; <u>atomic</u> <u>oxygen is transported from the surface to</u> the deeper region and (re)forms H₂O and HDO ices. " - Furuya et al 2013

High temperatures

Scenario 1: Mixing high D/H water up, reducing D/H (H₂O)

HDO H₂O

Low temperatures, Hatoms?

High temperatures

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HDO

Low temperatures, Hatoms?

Scenario 2: Mixing oxygen to the cold midplane, enhancing D/H (H₂O)



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High temperatures HDO H20

Low temperatures, Hatoms?

3) The role of mixing/turbulent evolution? * Need more constraints on disk turbulence, e.g., Hughes+2011, Guilloteau+2012, Teague+2015, Flaherty+2015 (α<9x10-4).



4) Differential Evolution of Solids and Ice

4) Differential Evolution of Solids

 Redistributes volatiles carried in the ices (Hogerheijde+2010, Bergin) +2016, Du+2015, 2016, sub.). Changes the C/O ratio (e.g., Piso+2015).



Vertical Settling



4) Differential Evolution of Solids







4) Differential Evolution of Solids TW Hya



Perhaps halted by substructure, formation of rings?

HL Tau



5) Baseline Models of Water and Simple Organic Deuteration: Variations in Deuteration Pathways

1) Key Ingredients for High Molecular HDO/H $_2O$



Supports fractionation up to T < 50 K

Starting out with $HDO/H_2O = HD/H_2$, how much does cold water formation in the disk contribute to elevated HDO/H_2O ?

The Classical Picture of Disk Ionization



Glassgold 1997, 2000, 2001 (and more), Igea & Glassgold 1999, Umebayashi+1989, 2009, Ilgner & Nelson 2006a/b, 2008.

An Updated Picture of Disk Ionization



Cleeves et al. 2015 measured a significantly subinterstellar CR ionization rate in TW Hya (> 100x reduced).

High densities + low ionization rates → very low ion fraction in the cold gas where deuterium enrichment is otherwise facilitated..

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Chemical Model

* Mini-deuterium chemical network designed to robustly predict HDO abundances.

* 6268 reactions, 600 species.

- * H₂/HD/D₂ self-shielding (Wolcott-Green+2011)
- * Simple grain-surface chemistry (Hasegawa, Herbst, Leung 1992),
- Thermal o/p ratios assumed for H₂ and * H_2D^+ (Lee & Bergin 2015)

* Warm fractionation reactions of Thi+2010.



And updated lab data on CO binding energies for oxygenregulation.

HDO/H₂O Results (1 Myr) H_3^+ and H_2D^+ 70 1.2 0.8 60 2e-5 0.4 9 50 0.0 2 3 5 4 \mathbf{O} 40 (AU) *D/H=6e-6* 30 1 m \mathbb{N} 20 -2 105 ()0 80 20 60 40 R (AU)





HDO/H2O Results (1 Myr)



HDO/H2O Results (1 Myr)



Initial Bulk Value Chemical Model at 1 Myr

at 1 Myr Earth's Oceans

Disk-Sourced Deuterium: Results

Chemistry in a laminar disk is <u>not</u> a viable source origin for HDO, H_2O or $D/H(H_2O)$ in the Solar System.

These conditions require ISM heritage such that interstellar ices would be incorporated into comets, meteorites, and Earth's oceans (30-40%).

But what else came along for the ride?

Cleeves, Bergin, Alexander, Du, Graninger, Öberg, Harries, 2014, Sci, 345, 1590.

D/H in Water vs. Organics Water H_2 Organics 10^{-1} Martian melt inclusions 10^{-2} Interstellar H₂CO Interstellar HCN Meteorites **Orgeuil Radical** Interstellar Ices D/H Ratio Lunar Apatite Hartley 2 HCN Interstellar CH₃OH 004 Q2 CH₄ ← CR IOM Hotspot e-Bopp HCN 10^{-3} VSMOW RAS 16293-2422 Env. CR IOM Bulk RAS 16293-2422 **Uranus** nterstellar H₂O Protosun C1/2 IOM Bulk Jupiter I 10^{-4} Comets Hal 10^{-5}

Cleeves, Bergin, Alexander, Du, Graninger, Öberg, Harries, 2016

D/H in Water vs. Organics

D/H Ratio

- * Globally higher organic D/H than water. Perhaps due to: $* CH_3^+ + HD \rightleftharpoons CH_2D^+ + H_2 + \Delta E$
 - * Operates in the forward direction even at 80-100 K due to higher ΔE~483-660 K (Roueff+2013).
- *Larger range in organic D/H = many reaction pathways?

Cleeves, Bergin, Alexander, Du, Graninger, Öberg, Harries, 2016



Organic Fractionation Pathways H_2^+ CH₂⁺ H_2 CH_2D^+ CH_3^+ H₂ H₃ H₂ CH_5^+ , CH_4D^+ H,D CO CO C(ice) $CH_3D + HCO^+$

$CH_4 + DCO^+$

 CH_4 , CH_3D , etc. (ice)



CH_3OH , CH_3OD , etc. (ice)

1) Roueff+2013, 2) Hugo+2009, 3) Adams&Smith 1985 Ilse Cleeves, CfA

Organic Fractionation Pathways



disk-like densities, 109-1012 cm⁻³

1) Roueff+2013, 2) Hugo+2009, 3) Adams&Smith 1985 Ilse Cleeves, CfA

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Cleeves, Bergin, Alexander, Du, Graninger, Öberg, Harries, 2016



Global Organic D/H



Now with 15,000 reactions, 1000 species

Global Organic D/H



Now with 15,000 reactions, 1000 species

Global Organic D/H



Now with 15,000 reactions, 1000 species

Midplane Organic Deuterium Fractionation



Orgueil Radicals IOM Hot-Spots



VSMOW



Midplane Organic Deuterium Fractionation



Orgueil Radicals IOM Hot-Spots



Ilse Cleeves, CfA

VSMOW

Interesting predictions for cometary D/H in CH3OH?

6) So what's happening?
T>500 K T<500 K Rcf ~O.1 Myr ~1 AU HDO H₂O











HDO/H₂O ~10⁻³

Rcf ~40 AU

HDOH₂O HDOH₂O





Rcf ~40 AU

~10 AU

H_2O HDO H_2O





~10 AU $HDOH_2O$ H_2O



~10 AU



Ingredients for Disk Deuterium Fractation





ALMA observations

But also many more puzzles!



$H^{13}CO^+ 3-2 \times 0.7$

Öberg et al 2015 Ilse Cleeves, CfA

Summary

- * The ionization, thermal, and chemical structure of protoplanetary disks impacts the deuterium chemistry, especially in the observable layers.
- * The assembly of the disk from the cloud can change the initial D/H affected by to cloud angular momentum and early accretion outbursts.
- * Mixing would be important for D/H (raising and lowering), but not clear if mixing is active in protoplanetary disks.
- * Transport of solids is observed and likely efficient carrier of volatile ices.
- * Organics more readily fractionated even in relatively warm gas due to high endothermicity of CH₂D⁺ + H₂ compared to H₂D⁺ + H₂.

